## **PHY 131**

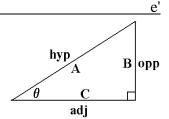
## THINGS TO KNOW

The following items were deliberately left off of the "Tables and Formulas" handout because you should know them.

## sec. 2:

Definitions of trig functions:

$$\sin\theta = \text{opp/hyp}, \quad \cos\theta = \text{adj/hyp}, \quad \tan\theta = \text{opp/adj}$$



## Pythagorean theorem: $A^2 = B^2 + C^2$

sec. 3:

Newton's 1<sup>st</sup>: If no net force, a body follows a straight line at a constant speed.

Newton's  $2^{nd}$ :  $\Sigma F = ma(\Sigma \text{ means summation.})$ 

Newton's 3<sup>rd</sup> (in sec 6, actually): If object A exerts a force on object B, then B exerts an equal and opposite force on A.

Weight: W = mg

sec. 5:

Kinetic energy:  $KE = \frac{1}{2}mv^2$ 

Gravitational potential energy:  $U_g = mgh$ 

Total mechanical energy: E = KE + U

Conservation of energy: Total energy (mechanical + all other forms) of an isolated system is constant.

<u>sec. 6</u>:

Momentum: p = mv

Conservation of momentum: Total momentum of an isolated system (no external forces) is constant. sec. 9:

Conservation of angular momentum: If no external torques, a system's total angular momentum is constant.

sec 10:

T = 1/f (T = period, f = frequency)

sec 11:

 $v = f\lambda$ ,  $f = v/\lambda$ ,  $\lambda = v/f$  (v = wave's speed, f = frequency,  $\lambda =$  wavelength

sec.12:

pressure: P = F/A (force per unit area)

Ideal gas law: PV = nRT

P = absolute pressure, V = volume, n = number of moles, R = gas constant, T = absolute temperature sec 14:

Definition of density:  $\rho = m/V$  (mass per unit volume)